

METHOD OF INHIBITING DEPOSITS IN COKE OVEN GAS PROCESSING EQUIPMENT

5 FIELD OF THE INVENTION

The present invention relates to a method of inhibiting deposits on the equipment and in the ducts and pipes carrying the flushing liquor used in the operation of coke ovens. More particularly, the present
10 invention relates to a method of inhibiting deposits in equipment handling flushing liquor in a coke oven system by treating the flushing liquor with quinoline in combination with 2-butoxyethanol.

BACKGROUND OF THE INVENTION

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In the operation of coke ovens which produce metallurgical coke from coking coal, deposits can form on the equipment and in the ducts and pipes which carry the coke oven gases. In processing and handling of the volatile coke oven gas evolved from the coal during the coking
20 process, deposits and fouling are common. The gases and vaporized liquids removed as effluent gas in a coking process can include tar, light oil, ammonia liquor etc. Some of the specific products refined from coke ovens include ammonium sulfate, benzene, toluene, xylene, naphthalene, pyridine, phenanthrene, anthracene, creosote, road tar, roofing pitches,
25 pipeline enamels, along with many other products. Several hundred

individual compounds have been found, organic and inorganic, in the effluent gas from a coking process.

A simplified description of the coking process would be the
5 destructive distillation of a complex carbonaceous mineral. The
compounds formed or driven off during the process have a wide range of
boiling and melting points and solubilities, causing the selective
condensation or crystallization of the higher boiling compounds. In
handling and processing the gases driven off during the coking process,
10 fouling of the handling and processing equipment often limits run time for
the equipment between shut downs for cleaning.

The coke oven gases from the coke oven are at high
temperatures, often about 800° C. These hot gases which leave the coke
15 oven by way of standpipes pass through goosenecks at the top of the
standpipes where the gases are cooled by flushing liquor. The flushing
liquor cools the gases from about 800° C to about 80° C so that the gases
can be safely handled by the downstream collector main and gas
processing equipment. The cooling also condenses tar and tar-mist
20 vapors which are then carried by the flushing liquor along the collector
main to a tar-liquor seal which directs the tar to decanters or
predecanters where tar is recovered. The flushing liquor also carries
solid coal, coke and cracked carbon particles from the gas stream and
into the tar to the decanters. The flushing liquor also dissolves nearly all
25 of the ammonium salts and much of the free ammonia thereby reducing
the contaminant levels in the coke oven gas.

The flushing liquor which passes through the tar-liquor seal is
"contaminated" by the multitude of organic and inorganic materials

present in the gases driven off during coking. The flushing liquor typically is treated in an ammonia recovery system. The presence of these contaminants leads to fouling of the goosenecks, collection mains and the ammonia recovery system with deposits.

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SUMMARY OF THE INVENTION

The present inventors have discovered that treatment of the flushing liquor with a heterocyclic nitrogenous compound in combination with 2-butoxyethanol can inhibit fouling deposits in the lines, conduits and equipment handling the flushing liquor thereby significantly extending run time between shut downs for cleaning. Addition of a heterocyclic nitrogenous compound such as quinoline in combination with 2-butoxyethanol such as Butyl Cellosolve® (Cellosolve is a registered trademark of Union Carbide Corporation) to the flushing liquor has been found to inhibit the formation of undesirable deposits on the lines, conduit and equipment which come into contact with the flushing liquor. It is also believed that the treatment combination, when added to the flushing liquor, can also result in removal of already formed deposits.

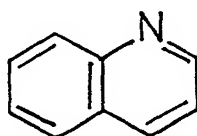
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DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

A method of inhibiting and dissolving deposits on conduits, lines and equipment handling flushing liquor in a coke oven plant is disclosed. The method comprises adding to flushing liquor in a coke oven plant, a heterocyclic nitrogenous compound and 2-butoxyethanol in an amount sufficient to inhibit and/or dissolve undesirable deposits. Preferably, the treatment combination is added to the flushing liquor prior to the

goosenecks at the top of the coke oven standpipes where the flushing liquor first contacts the coke oven effluent gases.

The heterocyclic nitrogenous compound of the treatment
5 combination of the present invention is preferably quinoline of the general formula:



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The 2-butoxyethanol compound of the treatment combination of the present invention is available as Butyl Cellosolve® from the Union Carbide Corporation. The ratio of heterocyclic nitrogenous compound to 2-butoxyethanol can range from about 50:50 to about 10:90 by weight
15 percent. The preferred ratio is about 25:75 weight percent.

The treatment combination is added to the flushing liquor in an amount sufficient to inhibit deposit formation in the conduits, lines and equipment which the flushing liquor comes into contact with. Typical
20 flushing liquor treatment combination concentrations can range from 0.5 to 5.0 parts per million.

In an operating coke plant, fouling of the conduits, lines and ammonia stills which contacted flushing liquor resulted in a short run life
25 between shut downs for cleaning of about three weeks. Upon addition of 5 parts per million of a combination of quinoline and Butyl Cellosolve® in a ration of 25:75 percent by weight to the flushing liquor to the ammonia still, the run life between shut downs for cleaning increased to several months.